

# Antibacterial Activitiy of Triterpeniod Compounds

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Abstract: The goal of the study was to ascertain whether particular pathogenic bacteria could be inhibited by triterpeniod substances obtained from Rosmarinus officinalis (rosemary) extracts. The time frame for completing this construction was July 2023-April 2024. At the beginning, 100 specimens were taken from patients who were undergoing treatment at the Burn Cure Center in Nineveh Governorate, Iraq Routine biochemical testing were used to determine burn infections caused by bacteria, and the Vitek2 compact system was used to confirm the diagnosis. Acinetobacter baumani and Pseudomonas auroginosa were the most common isolates. In order to extract the substance, rosemary (R. officinalis) leaves were air-dried in the shade and then processed into a fine powder using a grinding machine. A soxhlet device was used to extract 50 g of powdered leaves for six hours at 40–60 °C using 250 ml of 75% petroleum ether. Afterwards, the mixture was dried out at 40°C in a rotating evaporator. After being weighed, the extract residue was stored until needed. Using gas chromatography/mass spectrometry (GC/MS), the crude extract was examined. Ursolic acid (1.38), oleanolic acid (3.27), and other triterpeniod substances with varying ratios were identified by GC-MS analysis. The antibacterial activity of triterpeniod compounds was assessed at concentrations of 12.5%, 25%, 50%, and 100% using the well-diffusion Kirby-Bauer method. The test was evaluated in comparison to six bacterial strains. The results indicated that staphylococcus aureus (50%:22mm - 100%: 26mm), staphylococcus haemolyticus (100%: 17mm), and Acinetobacter baumani (50%:18mm – 100%: 22mm) had the highest inhibitory.

Keywords: Rosmarinus Officinal is, Triterpeniod, Antibacterial Activity, GC-MS.

## 1. INTRODUCTION

The skin, being the largest organ in the body, is considered to be one of the most vital barriers against external infections (Dąbrowska, A. K., et al., 2018). This organ also plays a role in sensory perception, hormone balancing, temperature and humidity management, and burn injuries that cause skin tissue deterioration and loss of integrity (Dąbrowska, A. K., et al,



2018). Resolving burn injuries is becoming more challenging, making it one of the most prevalent public health emergencies globally. Burn injuries are caused by electric shock, radiation, fire, strong acids, fuel, and other toxic substances. They can also result from organic tissue injury (Uddin et al., 2018). Because burns remove the skin's outer layer, bacteria can penetrate and infect the injured area (Mesbahi et al., 2021)

Burn injuries and the ensuing infections have been one of the main issues facing medical systems worldwide (Kelly et al., 2022). An estimated 38,000 people in the US sought medical assistance and were admitted to burn centers in 2018 according to reports (Glat et al., 2024). Burn infections continue to be one of the leading causes of death in the modern world, despite the advancements in burn care over the past 50 years (Vinaik et al., 2019). Many studies over the past ten years have shown that 42-65% of burn patients pass away from microbial infections (Vinaik et al., 2019). Furthermore, those with infectious burns have a death rate that is about twice as high as people with non-infectious burns (Medisa Primasari & Budi, 2024). The removal of the skin's protective layer, which impairs immunity and makes the body more vulnerable to infections, is the most detrimental effect of burns (Gour. et al, 2024). Moreover, the more complex interaction of antiinflammatory signals results in anomalies in the innate and adaptive immune systems (Gour. et al., 2024). Furthermore, tracheal intubation, arterial lines, venous and urinary catheters, and prolonged hospital stays have been associated with an increased risk in patients with burns (Duan et al., 2024). Antibiotic resistance emerged rapidly after the first antibiotic was discovered and is presently one of the largest issues facing the medical community, especially in burn units (Liu et al., 2024). In addition to posing a serious risk of nosocomial infections, including lung, urinary tract, and cellulitis infections, multidrug-resistant microbes are one of the main reasons of death for burn patients. The development of multidrug-resistant illnesses is undoubtedly influenced by both empirical antibiotic therapy and extended hospital stays. Gram-positive bacteria were the most often isolated pathogens in the early days of admitting burn patients, and their antibiotic profile was more sensitive. With time, gram-negative bacteria develop greater resistance. (Hemmati et al, 2023)

Aromatic rosemary is a native of the Mediterranean region, Rosmarinus officinalis L. (Labiaceae). Rosemary essential oils (REOs) have been utilized in aromatherapy, preparing food, and medical applications since antiquity (Meziane. et al, 2024), and they are recognized in many nations and listed in official pharmacopeias ( Lolas .et al,2024). The antibacterial, insecticidal, anti-inflammatory, and antioxidant qualities of these substances have led to their widespread application in cosmetics, therapy, and sterilizing (González-Minero et al., 2020). Furthermore, these potentially useful organic substances have been applied in novel ways in other fields, including medicine delivery methods (Kalaki Kordkolaei et al., 2020), nonantibiotic feed additives (Sgarro et al., 2024), and innovative packaging techniques (Hosseini et al., 2021). REOs are used as flavorings in food, packaging, and skin care products for a long time in the cosmetics industry.

## 2. RELATED WORKS

Though much remains unknown about their pharmacology, pentacyclic triterpenoids are among the most important compounds identified in plants. The triterpenoids with oleanane,



ursane, and lupane skeletons—most notably oleanolic acid and ursolic acid—are the most often studied of these secondary plant metabolites.Gudoityte et al., 2021).

In many scientific and technological domains, gas chromatography (GC) is a widely employed method. GC has been important in determining the number of components and their ratios in a combination for more than 50 years. It is uncertain and constrained how these separated and measured molecules' types and chemical structures can be determined, which calls for the use of a spectrophotometer detecting tool. Of them, the mass spectrometric detector is the most commonly used. (MSD), which offers the mass spectrum—or the molecule's "fingerprint" When a high resolution mass spectrometer is used, mass spectral examinations provide the molecular weight, element composition, presence of functional groups, and, in certain cases, the structure and location isomerism of the molecule.(Muhamma, al., 2024)

Mass spectrometry and gas chromatography used to detect different chemicals in an extract is known as gas chromatography/mass spectrometry (GC/MS).( Muhammad et al., 2024). One study goal was to include: Using the GC-Mass method, extract the triterpeniod components from the rosemary plant (Rosemarinus officinalis), and investigate the antibacterial activity of the rosemary extract against certain isolates.

## 3. METHODOLOGY

#### Materials and Methods:

#### **Collection of Burns Samples:**

This construction was completed between July 2023 and April 2024. Initially, patients at the Burn Cure Center in Nineveh Governorate, Iraq, provided one hundred specimens. Based on standard biochemical testing, bacteria isolated from burn infections were identified, and the Vitek2 compact system was used to validate the diagnosis.

#### Solutions:

## 1. The McFarland Turbidity Standard

The McFarland No. 0.5 turbidity standard before standardizing the amount of bacterial cells, which should be  $1.5 \times 108$  CFU/ml, the components were well combined.

2. Sigma-Aldrich (Sigma-Aldrich, Darmstadt, Germany) was the source of the analytical grade dimethyl sulfoxide (DMSO).

#### Plant Samples Collection, Preparing, and Extraction of Essential Oils:

After being gathered from various neighborhood stores in Mosul, the rosemary (R. officinalis) leaves were air-dried in the shade and ground into a fine powder using a grinding machine. Using a soxhlet apparatus, 50 g of leaf powder were extracted over the course of six hours at 40–60 °C using 250 ml of 75% petroleum ether. A rotary evaporator operating at 40°C was then used to dry the solution. For subsequent research, 400 mg/ml of rosemary extract was reconstituted in dimethyl sulfoxide (DMSO) (Jai Kumar & Geetha, 2021).

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# Gas Chromatography/Mass Spectrometry (GC-MS):

The components of R. officinalis' extracted essential oils were examined using a Perkin Elmer Clarus 500 gas chromatograph connected to a Perkin Elmer Clarus 560 mass spectrometer. A Perkin Elmer Elite-5 fused-silica capillary column (30 m × 0.25 mm, film thickness 0.25  $\mu$ m) was employed to separate the components of the EOs. The temperature of the column was set to vary at a rate of 4 °C/min from 50 °C for 5 min to 280 °C. Throughout all of the chromatographic runs, the carrier gas flow rate of helium was maintained at 1 mL/min. 0.2  $\mu$ L of pure EO was injected in split mode at 250 °C with a splitting ratio of 1:50. A comprehensive scan mode encompassing 50-500 m/z was gathered. By comparing the retention indices of the EOs to those documented in the literature and their MS to reference spectra in the NIST mass spectrometry data center, the chemical components of the EOs were identified.

## 4. RESULTS AND DISCUSSION

## **Results:**

The identification and quantification of thirty distinct compounds from R. officinalis EOs were made possible by the GC-MS analysis results (Figures 1 and 2). These substances, which are triterpeniod compounds, such as ursolic acid (1.38) and oleanolic



Acid (3.27) Figure (1) Data Path: D:\MassHunter\GCMS\1\data\ Data File: ROSMARINUS OFFICINALIS Acq On: 17 Dec 2023 10:20 Operator: Sample: ROSMARINUS OFFICINALIS. Misc: ALS Vial: 1 Sample Multiplier: 1 Search Libraries: C:\GCMS\firmware\NIST11.L

| Peak | <b>Retention Time</b> | Area% (Concentrations) | Library/ Id                          | Reference |  |
|------|-----------------------|------------------------|--------------------------------------|-----------|--|
| 26   | 29.165                | 1.38                   | C:/GCMS/<br>firmware<br>Ursolic acid | 128345    |  |

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|    |        |      | C:/GCMS/       |        |
|----|--------|------|----------------|--------|
| 27 | 29.390 | 3.27 | firmware       | 181285 |
|    |        |      | Oleanolic acid |        |

Figure -2

The antibacterial activity:

The antibacterial activity of triterpeniod compounds were assessed against six bacterial strains (at concenttrations 12.5%, 25%, 50%, 100%)

| strains                     | 12.5 % | 25 % | 50% | 100 % |
|-----------------------------|--------|------|-----|-------|
| Acinetobacter baumani       | 16     | 16   | 18  | 22    |
| Staphylococcus aureus       | 15     | 19   | 22  | 26    |
| Staphylococcus haemolyticus | R      | R    | 13  | 17    |
| E.coli                      | R      | R    | R   | R     |
| Pseudomomonas auroginosa    | R      | R    | R   | R     |
| Proteus mirabilis           | R      | R    | R   | R     |

Figure (3)

## **Discussion:**

The findings illustrated in Figure 3 indicate that the triterpeniod compounds exhibited greater efficacy against staphylococcus aureus (gram-positive bacteria) in comparison to Acinetobacer baumani (gram-negative bacteria). The structure of these bacterial types' cell walls could be the cause of this. Moreover, efflux pumps, which remove several substances from the periplasm and transport them outside of the cell, strengthen the intrinsic resistance of gram-negative bacteria (Maillard & Pascoe, 2024).

## 5. CONCLUSION

The results of this study show that the triterpeniod compounds exhibited potent antibacterial properties, suggesting that they could be employed as both natural and pharmaceutical treatments to treat bacterial infections that cause burns.

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